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	作成者: SATO, Kazuhiro, KUSAKA, Yukinori, OKADA,
	Kenichiro
	メールアドレス:
	所属:
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Direct effect of cadmium on citrate uptake by isolated rat renal brush border membrane vesicles

Kazuhiro Sato*^a, Yukinori Kusaka^a, Kenichiro Okada^b

^aDepartment of Environmental Health, Fukui Medical School, 23 Shimoaizuki, Matsuoka-cho, Fukui 910-11, Japan ^bDepartment of Urology, Fukui Medical School, 23 Shimoaizuki, Matsuoka-cho, Fukui 910-11, Japan

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Abstract

High incidence of multiple kidney stone formation has been observed among workers exposed to cadmium (Cd). Citrate is known to be a protective factor against renal stone formation. To study the direct effect of cadmium on citrate uptake by the renal brush border membrane, we exposed isolated rat renal brush border membrane vesicles (BBMV) to cadmium and determined their citrate uptake characteristics. BBMV were prepared by the divalent cation precipitation method. Citrate uptake was measured by the Millipore rapid membrane filtration technique. Preincubation of BBMV with 2 and 10 mM CdCl₂ for 1 min significantly inhibited citrate uptake compared with that of BBMV without Cd. Analysis of the time course of citrate uptake during 30-min preincubation of BBMV with 0.5 mM Cd also revealed significant reduction of the uptake compared with that of the control BBMV without preincubation. These findings indicate that preincubation of BBMV with cadmium results in time-dependent and concentration-dependent inhibition of citrate uptake.

Keywords: Cadmium; Renal stone; Citrate; Brush border membrane vesicle

1. Introduction

The incidence of urolithiasis in Japan, like that in other developed countries, has been steadily increasing since the Second World War [1], although the etiology of this condition is not completely understood. Occupational exposure to cadmium is associated with renal damage, particularly with proximal renal tubular dysfunction [2,3]. Chronic exposure to cadmium produces a renal tubular syndrome which includes aminoaciduria, proteinuria and hypercalciuria [4]. Some epidemiological studies have revealed that workers exposed to cadmium have a high incidence of renal stones [2]. Cadmium inhibits the Na-D-glucose [5], Lalanine [6] and L-glutamate [7] co-transport system in the rat renal brush border membrane. On the other hand, citrate is a well known inhibitor of the formation of calcium-containing renal stones, and has been shown to reduce the saturation of stone-forming calcium salts by forming a soluble complex with calcium [8]. The mechanism by

^{*} Corresponding author.

which cadmium affects the Na-citrate co-transport system, however, has not been clarified.

In the present study, we exposed rat renal brush border membrane vesicles (BBMV) to cadmium and investigated the direct effect of this exposure on their citrate uptake. The results presented here suggest that cadmium might inhibit the Na-citrate co-transport system in renal brush border membranes.

2. Materials and methods

2.1. Materials

¹⁴C-Citrate (spec. act. 50 mCi/m mol) was obtained from the New England Nuclear Corp. (Boston, MA). All other chemicals were of reagent grade and purchased from commercial sources.

2.2. Animals

Experiments were performed using isolated brush border membranes obtained from male Wistar rats (CLEA, Japan, Inc.) weighing 290-340 g. They were fed a control diet and tap water ad libitum in a temperature and humidity-controlled room under a 12-h light-dark cycle.

2.3. Brush border membrane isolation

Each brush border membrane specimen was prepared from the cortex of both kidneys of a single rat by a modification of the MgCl₂-precipitation technique [9]. During preparation, all materials were maintained at 4°C. Briefly, each rat was anesthetized with chloral hydrate (360 mg/kg body wt.) administered intraperitoneally. The renal cortices were removed and placed in isolation buffer consisting of (in mM) 50 mannitol and 2 Tris-(hydroxymethyl)-aminomethane (Tris)/HCl at pH 7.0. They were homogenized in a glass Teflon homogenizer (Iuchi Co. Ltd., Japan) and a polytron homogenizer. A concentration of MgCl₂ (1 M) was added to the resultant homogenate to give a final concentration of 10 mM. This mixture was stirred for 20 min in a cold room and then centrifuged for 15 min at 3000 \times g. The supernatant was collected and centrifuged for 20 min at 43 000 \times g. The pellet containing the brush border membrane vesicles was resuspended in isolation buffer and centrifuged for 15 min at $3000 \times g$. The supernatant was centrifuged for 20 min at 43 000 \times g. The pellet containing the purified brush border membrane vesicles (BBMV) was resuspended in (in mM) 260 mannitol and 20 N²-hydroxyethyl-piperazine-N'2-ethanesulfonic acid (Hepes)/Tris, pH 7.0, at a final concentration of about 10 mg protein/ml.

2.4. Enzyme assays

The purity of the membrane preparations was assessed by evaluation of the enrichment of alkaline phosphatase (ALP) [10] and leucine aminopeptidase (LAP) [11] in the final preparation as compared with that of original homogenate.

2.5. Protein determination

Protein determination was conducted by the method of Lowry et al. [12] using crystalline bovine serum albumin as the standard.

2.6. Uptake studies

Citrate uptake is sodium dependent and electrogenic such that positive charge is transported [13]. The uptake of citrate by the BBMV was determined by the Millipore rapid membrane filtration technique [14-17]. In time course experiments, membrane vesicles (10 μ l) were incubated at 30°C with 40 μ l of uptake solution consisting of (in mM) 0.1 ¹⁴C-citrate, 130 NaCl or KCl, 20 Hepes/Tris, with or without 0.5 CdCl₂, pH 7.0. At appropriate intervals, the uptake was terminated by the addition of 4 ml of ice-cold solution consisting of (in mM) 130 NaCl or KCl, 20 Hepes/ Tris, pH 7.0. The solution was then rapidly filtered through a Millipore filter (DAWP02500; pore size, 0.65 μ m) and washed twice with 4 ml of ice-cold stop solution under vacuum suction. For the experiment in which BBMV were preincubated with CdCl₂, membrane vesicles (10 μ l) were preincubated at 30°C with 10 µl of preincubation solution consisting of (in mM) 260 mannitol, 20 Hepes/Tris, pH 7.0, with or without CdCl₂. Citrate uptake was initiated by the addition of 40 μ l of uptake solution consisting of (in mM) 0.1 ¹⁴Ccitrate, 130 NaCl, 20 Hepes/Tris, pH 7.0. After thirty seconds, uptake was terminated by the addition of 4 ml of ice-cold solution consisting of (in mM) 130 NaCl, 20 Hepes/Tris, pH 7.0. The solu-

1000

500

CITRATE UPTAKE (p mol/mg protein)

tion was then rapidly filtered through a Millipore filter (DAWP02500: pore size 0.65 μ m) and washed twice with 4 ml of ice-cold stop solution under vacuum suction. Background uptake was determined similarly by adding 4 ml ice-cold stop solution to membrane vesicles (10 μ l) before adding 40 μ l of uptake solution at 0°C. Non-specific binding was always < 0.5% of the total counts in the uptake solution. Radioactivity associated with the filters was measured with a liquid scintillation counter (LSC-3500, Aloka). All uptake measurements were performed at 30°C in triplicate, and uptake was calculated on the basis of specific activity measured in each experiment. The value of non-specific binding was subtracted from the experimental value and the vesicular uptake was expressed as picomoles ¹⁴C-citrate per mg protein. Student's t-test was used to analyze differences among the groups. Significance was accepted at P < 0.05.

3. Results

The purity of the BBMV preparations was assayed by determining the BBM-specific marker enzymes ALP and LAP. The specific activities of the two enzymes in the BBM fraction were enriched by about 10-fold over those in the original homogenate.

3.1. Effect of cadmium in the uptake solution on time course of citrate uptake by BBMV

When BBMV were prepared in NaCl-free medium and incubated in 130 mM NaCl-containing buffer, the uptake of citrate showed a characteristic 'overshoot' phenomenon during the first 2 min and reached equilibrium after 120 min (Fig. 1). The overshoot indicates an intravesicular accumulation of citrate above the equilibrium concentration and occurs because of the persistence of a Na⁺ gradient when the intravesicular citrate has already reached the concentration in the incubation medium [18]. But there was no significant difference between the uptake solution with 0.5 mM $CdCl_2$ and the uptake solution without $CdCl_2$. When the Na⁺ gradient was replaced by a K⁺ gradient, the initial uptake of citrate was very low and the overshoot did not occur (Fig. 1).



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3.2. Effect of cadmium and cadmium concentration on citrate uptake by BBMV

The effect of preincubation of BBMV with 0-10 mM Cd for 1 min at 30°C is illustrated in Fig. 2.



Fig. 2. Effect of preincubation for 1 min (30°C) with cadmium at various concentrations on citrate uptake by brush border membrane vesicles. Values are means \pm S.E. **P* < 0.002 vs. uptake at 0 mM.

r

4

Cont. Na⁺ – gradient

Cont. K⁺ – gradient

к+

• Cd

▲ Cd

Na⁺ – gradient

gradient



Fig. 3. Effect of cadmium on citrate uptake by brush border membrane vesicles with respect to pre-incubation time. Values are means \pm S.E. *P < 0. 02 vs. 0 min.

Preincubation with 2 and 10 mM Cd significantly inhibited the citrate uptake compared with that without preincubation. These findings indicate that preincubation with Cd resulted in concentration-dependent inhibition of citrate uptake by BBMV.

3.3. Effect of cadmium on citrate uptake by **BBMV** with respect to preincubation time

Fig. 3 shows the time course of inhibition of citrate uptake after preincubation with 0.5 mM Cd. Following preincubation for 1 and 10 min, citrate uptake by BBMV was decreased, but not significantly so. Following 30-min preincubation, significant inhibition of the citrate uptake compared with that of the control without preincubation was observed. These findings indicate that Cd preincubation resulted in time- and concentration-dependent inhibition of citrate uptake by BBMV.

4. Discussion

The kidney is a major target organ of cadmium in animals and humans [4]. Cadmium impairs various components of the transport system in the proximal tubules [5–7]. Proteinuria is often an early sign of cadmium intoxication, and changes in reabsorption of calcium also occur early in the course, leading to hypercalciuria. In cadmium intoxication, hypercalciuria without proteinuria is sometimes observed [2]. The results of a previous epidemiological study revealed that the prevalence of renal stones in non-proteinuric men was higher than that in proteinuric men, suggesting that more developed tubular dysfunction was associated with lower risk of development of kidney stones [2]. However, the mechanism by which cadmium affects the transtubular transport process in the kidney is not known.

Na-dependent D-glucose uptake by BBMV is inhibited by Cd because Cd binds the external site of BBMV and reacts with sites on the Na-Dglucose co-transporter [5]. In the time course experiment, there was no significant difference between the uptake solution with and without 0.5 mM CdCl₂. However, preincubation of BBMV with cadmium resulted in time- and concentrationdependent inhibition of the citrate uptake by the BBMV. So, citrate is thought to have been taken up by the BBMV before Cd binding could occur.

Workers exposed to cadmium for a long period or at high dose, therefore, might have increased urinary citrate excretion, which prevents renal stones. In contrast, short-term or low-dose exposure to cadmium may cause hypercalciuria but may not increase urinary citrate excretion, which may cause a predisposition to renal stones. Urinary citrate excretion is extremely sensitive to changes in the systemic acid-base balance [19] and in potassium balance [17]; excretion is increased in metabolic alkalosis and decreased in metabolic acidosis and potassium depletion, the latter of which is associated with renal intracellular acidosis [20]. Cadmium intoxication may lead to metabolic and respiratory acidosis [2]. The urinary citrate excretion in individuals with cadmium intoxication seems to be determined by the two factors of acidosis and impairment of the citrate transport system in the brush border membrane.

In conclusion, we infer from epidemiological studies that individuals with cadmium intoxication and hypercalciuria have a high risk of developing renal stones. Nevertheless, the findings of preincubation of rat BBMV with cadmium indicated that high direct exposure to cadmium impaired the citrate uptake by the BBMV and might reduce risk of development of renal stones. In vivo experiments (cadmium-intoxicated rats) are, thus, needed to study a possible mechanism for renal stones observed among Cd-exposed workers.

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